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Note

High-performance liquid chromatographic determination of metrizamide in plasma

M. PETER KULLBERG, CHARLES E. BIDDLECOME, RICHARD W. ROSS and JEROME EDELSON

Department of Drug Metabolism and Disposition, Sterling-Winthrop Research Institute, Rensselaer, N.Y. 12144 (U.S.A.)

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Metrizamide, 2-[[3-(acetylamino)-5-(acetylmethylamino)-2,4,6-triiodobenzoyl]amino]-2-deoxy-D-glucose, Amipaque[®], is a non-ionic, water-soluble radiographic contrast medium intended for lumbar, thoracic and cervical myelography, ventriculography and cisternography. The absorption of metrizamide from cerebrospinal fluid to blood has been reported¹ using a technique based on the analysis of the specimen for its iodine content by a non-specific ashing technique².

This paper describes the development of a specific high-performance liquid chromatographic (HPLC) method for metrizamide in plasma. This assay technique has been used to measure quantitatively metrizamide in the plasma of human subjects undergoing radiographic evaluation in subarachnoid diagnostic applications.

EXPERIMENTAL

Materials

Metrizamide is a product of Winthrop Laboratories (New York, N.Y., U.S.A.). The internal standard for the assay, 3-dimethylaminomethyleneimino-2,4,6-triiodobenzoic acid, was synthesized at the Sterling-Winthrop Research Institute. The column was purchased from Whatman (Clifton, N.J., U.S.A.). Other chemicals were obtained commercially (reagent grade) and used without further purification.

Preparation of samples

Plasma standards were prepared by supplementing 1.0 ml of control human plasma with aliquots of an aqueous stock solution of metrizamide (1 mg/ml). Duplicate standards at concentrations of 0, 2, 4, 8, 12, 16 and 24 $\mu\text{g}/\text{ml}$ of metrizamide were prepared.

Two sets of randomized and coded plasma samples, to be analyzed under single blind conditions, were prepared as described above. Each set contained quadruplicate plasma samples at final concentrations of 0, 2.5, 5.5, 9, 15 and 19 $\mu\text{g}/\text{ml}$ in control human plasma. One set of samples was analyzed immediately and the other set was analyzed following a 5-day storage period in a -4° freezer. Freshly prepared

plasma standards were extracted and analyzed concomitantly with each set of unknown samples.

Protein precipitation and analysis

To 1.0 ml of human plasma, containing the disodium salt of N,N'-1,2-ethanediybis-[N-(carboxymethyl)glycine] (EDTA), as the anticoagulant, was added 50 μ l internal standard solution (1.5 mg/ml) and 0.2 ml of zinc sulfate solution (20%, w/v in water). The sample was mixed and 0.2 ml of a saturated aqueous solution of barium hydroxide was added. The tube was again mixed and a marble was placed over the opening of the test tube which was heated on a steam bath for 2 min. The tube was cooled and centrifuged.

A 25- μ l aliquot of the clear supernatant solution was analyzed under the following HPLC conditions. Column: Partisil 10 ODS (250 mm \times 4.6 mm I.D.) with a Corasil C₁₈ precolumn (60 mm \times 4 mm I.D.). Detector: Schoeffel Model SF 770 variable wavelength UV detector set at 244 nm. Mobile phase: distilled water-methyl alcohol-glacial acetic acid (94.5:5:0.5). Flow-rate: 2.0 ml/min. Retention time: metrizamide, 3.9 min; internal standard, 11.2 min. Pump: Milton Roy Model 709, 1000 p.s.i. Temperature: 20°, ambient.

Clinical study

Eight patients received a single injection of metrizamide in the subarachnoid space as part of a routine neuroradiological procedure. Blood specimens were obtained at 0, 2, 4, 6, 8, 24 and 48 h after administration. These were collected in vacuum tubes containing EDTA as the anticoagulant. The samples were centrifuged and the plasmas were separated and frozen until the analysis for metrizamide was performed.

RESULTS AND DISCUSSION

Recorder tracings from the HPLC analysis of both a control plasma sample and a sample supplemented with 16 μ g/ml of metrizamide are shown in Fig. 1.

A linear regression model was used to describe the relationship between the peak-height ratio of metrizamide to the internal standard and the concentration of

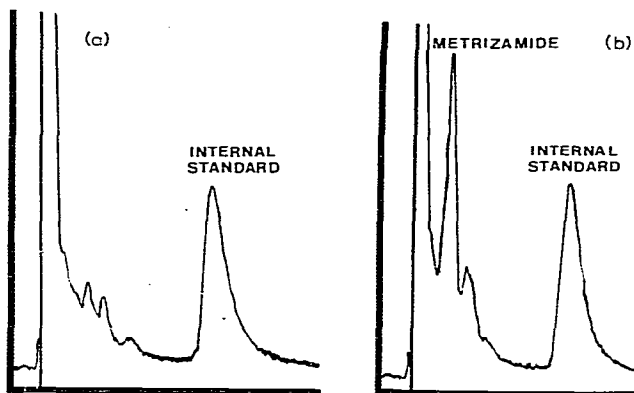


Fig. 1. Chromatogram of control human plasma containing the internal standard (a) and the same sample containing 16 μ g/ml of metrizamide (b).

metrizamide in the plasma standards. To estimate assay sensitivity, accuracy, precision, and to determine whether metrizamide concentration changes upon storage of frozen plasma samples, identical sets of plasma samples were assayed immediately and after storage; the plasma parameters were in good agreement. The minimum quantifiable level was determined by inverse prediction³ as that concentration whose lower 80% confidence limit just encompasses zero⁴. The overall mean sensitivity for metrizamide in plasma was 0.72 $\mu\text{g}/\text{ml}$. A representative plot of the relationship between relative peak height and the concentration of metrizamide added to plasma, is presented in Fig. 2.

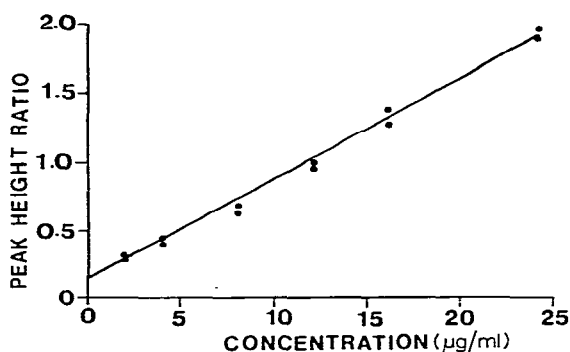


Fig. 2. Extracted standard curve of control human plasma augmented with metrizamide; each concentration was prepared and chromatographed in duplicate.

For the range of concentrations used in the single blind study, results were precise, accurate and reproducible. Table I summarizes the results for the analysis of the "unknown" plasma concentrations. There was no statistically significant difference among concentration groups with respect to mean percent differences. Over the range of these samples the method has an estimated precision of 5.1% (standard deviation). The estimated accuracy ranged from 3% low to 10% high over the concentration range. A logarithmic transformation was applied to the plasma data prior to statistical analysis to minimize the observed correlation between mean response and variance of the quadruplicate sets⁵.

The mean concentrations in the plasma of subjects at intervals after subarachnoid administration of metrizamide is shown in Fig. 3. Although the dose varied from 2.0 to 3.2 g of total iodine (4.1 to 6.6 g metrizamide), when the mean plasma concentrations are plotted on a semi-logarithmic scale against time, a linear relationship appears. This suggests first-order kinetics with an apparent half-life of about 11 h. Golman¹, using mean values of his measurements, has reported that "the biological half-life of metrizamide in the subarachnoid space was found to be 11 h". The mean rate constant for excretion of metrizamide was reported as 0.09 ± 0.01 (ref. 1) which corresponds to a half-life of 7.7 h; the range of reported values was from 0.23 (half-life of 4 h) to 0.04 (half-life of 17.3 h). The agreement with our estimated half-life in plasma suggests that the rate-limiting step in the excretion of metrizamide is the transfer from the subarachnoid space into the plasma.

TABLE I

CONCENTRATIONS FOUND IN PLASMA SAMPLES SUPPLEMENTED WITH METRIZ-AMIDE

Mean % difference = Mean of $[\text{antilog}(\log_{10} \text{ assayed} - \log_{10} \text{ seeded}) - 1] \times 100$. MQL = minimum quantifiable level.

Seeded levels ($\mu\text{g/ml}$)	Assayed level ($\mu\text{g/ml}$)	
	Day 1	Day 2
0	<MQL	<MQL
0	<MQL	<MQL
0	<MQL	<MQL
0	<MQL	<MQL
2.50	2.55	2.61
2.50	2.37	2.71
2.50	2.46	2.89
2.50	2.34	2.81
Geometric mean	2.43	2.75
% S.E.	1.96	2.24
Mean % difference	-2.9	+10.1
5.50	5.39	5.41
5.50	5.31	5.51
5.50	5.83	5.45
5.50	6.12	4.98
Geometric mean	5.65	5.33
% S.E.	3.40	2.34
Mean % difference	+2.8	-3.0
9.0	9.04	9.38
9.00	9.65	9.37
9.00	9.40	9.38
9.00	10.3	11.1
Geometric mean	9.59	9.78
% S.E.	2.80	4.33
Mean % difference	+6.5	+8.7
15.0	17.2	16.6
15.0	15.5	15.7
15.0	15.9	16.4
15.0	15.7	16.5
Geometric mean	16.1	16.3
% S.E.	2.37	1.27
Mean % difference	+7.1	+8.6
19.0	20.3	20.9
19.0	20.9	21.2
19.0	20.3	19.7
19.0	21.3	19.8
Geometric mean	20.7	20.4
% S.E.	1.19	1.88
Mean % difference	+8.9	+7.3

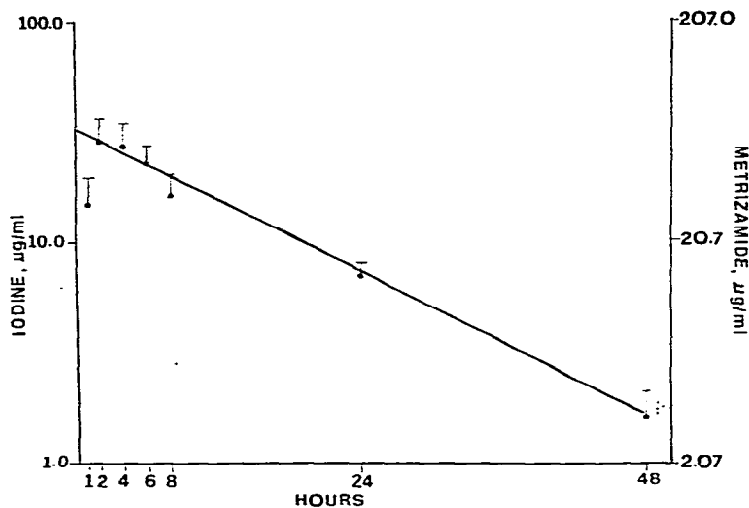


Fig. 3. Mean concentration of metrizamide in plasma of human subjects. Vertical lines denote 1 S.E. ($n = 8$).

In summary, an accurate, specific, reproducible and precise HPLC assay has been developed for the measurement of metrizamide concentration in human plasma. This method permitted estimation of the apparent first-order terminal elimination half-life in the plasma of human subjects that received metrizamide as a diagnostic agent.

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